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Diethyl 3-(1-Morpholinyl)-2,8-cyclooctadiene-1.2-dicarboxylate

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Abstract

The 2,8-cyclooctadiene moiety of the title compound, $C_{18}H_{27}NO_5$, has a *cis,cis* configuration. The morpholinyl ring adopts an almost ideal chair conformation.

Comment

Enamines comprise a group of useful intermediates in several organic syntheses (Cook, 1969). Reactions of enamines of cyclic ketones with activated acetylenes is a suitable method for ring enlargement (Lown & Akhtar, 1973; Reinhoudt & Kouwenhoven, 1972). Different products can be obtained by changing the type of cyclic enamine and the acetylene (Özbey, Temel, Tunoğlu & Uludağ, 1996). The structure determination of the title compound, (I), was undertaken in order to establish the configuration of the octadiene moiety and to compare the values obtained with reported structural results.



The structural results show that the title molecule has a cis, cis configuration of the 1,3-diene moiety. The torsion angles for the C=C-C=C part of the eightmembered ring are C7-C8-C1-C2 3.0(5), C8-C1—C2—C3 42.9(4) and C1—C2—C3—C4 $24.4(4)^{\circ}$. These values and the bond distances in Table 1 show that there is some conjugation between the double bonds in the cis, cis-1, 3-diene moiety. The conformation of the eight-membered ring is similar to that of dimethyl cis, cis-7, 8-dihydro-6-(1-pyrrolidinyl)-2Hthiocine-4.5-dicarboxylate (Reinhoudt et al., 1984) in which the corresponding torsion angles for the double bonds are -3.7(2) and $-17.4(2)^{\circ}$, and the double-bond lengths are 1.340(3) and 1.386(2) Å. In the present D_m not measured

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compound, the carboxylate groups attached to the C2 and C1 atoms are oriented with torsion angles C3-C2-C9-O2 and C8-C1-C12-O4 of -159.4(3)and 32.1 (4)°, respectively. The morpholinyl ring has an almost ideal chair conformation. The perpendicular distances of the two chair atoms in the para positions N1 and O5 from the plane of the other four atoms of the six-membered morpholinyl ring are -0.580(3) and 0.659 (3) Å, respectively. The torsion angle C4-C3-N1-C15 is $-162.1(3)^{\circ}$.



Fig. 1. The molecular structure of C18H27NO5 with the atomnumbering scheme. The displacement ellipsoids are drawn at the 50% probability level and H atoms are drawn as small spheres of arbitrary radii.

Experimental

A solution of diethyl acetylenedicarboxylate (5.46 mmol in 15 ml benzene) was added over a period of 2 h to a solution of morpholinylcyclohexene (5.46 mmol in 10 ml benzene) at 278-280 K to yield the title compound. The reaction mixture was stirred for 6 h after which the solvent was removed under reduced pressure. Column chromatography of the residue [ethyl acetate-benzene (1:4)] afforded the pure product (yield 56%; m.p. 427 K).

Crystal data

-	
C ₁₈ H ₂₇ NO ₅	Mo $K\alpha$ radiation
$M_r = 337.42$	$\lambda = 0.71073 \text{ Å}$
Triclinic	Cell parameters from 25
PĪ	reflections
a = 8.6903 (12) Å	$\theta = 8 - 18^{\circ}$
b = 10.3999 (14) Å	$\mu = 0.091 \text{ mm}^{-1}$
c = 11.9025(13) Å	T = 295 K
$\alpha = 68.481 (1)^{\circ}$	Prismatic
$\beta = 71.813(1)^{\circ}$	$0.56 \times 0.24 \times 0.16$ mm
$\gamma = 64.844 (1)^{\circ}$	Colourless
$V = 889.98 (15) \text{ Å}^3$	
<i>Z</i> = 2	
$D_x = 1.259 \text{ Mg m}^{-3}$	
D not macquired	

Data collection

Enraf–Nonius CAD-4	$R_{\rm int} = 0.02$
diffractometer	$\theta_{\rm max} = 26.3^{\circ}$
$\omega/2\theta$ scans	$h = -10 \rightarrow 9$
Absorption correction: none	$k = -12 \rightarrow 0$
3808 measured reflections	$l = -14 \rightarrow 13$
3392 independent reflections	3 standard reflections
1805 reflections with	frequency: 120 min
$I > \sigma(I)$	intensity decay: 0.2%

Refinement

Refinement on F	$(\Delta/\sigma)_{\rm max}$
R = 0.0401	$\Delta \rho_{\rm max} = 0$
wR = 0.0419	$\Delta \rho_{\rm min} = -$
S = 1.26	Extinction
1805 reflections	Scattering
217 parameters	national
H atoms: see below	Crystall
$w = 1/[\sigma^2(F^2) + (0.04F^2)^2]$	-

 $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.241 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -0.116 \text{ e } \text{\AA}^{-3}$ Extinction correction: none Scattering factors from International Tables for X-ray Crystallography (Vol. IV)

Table 1. Selected geometric parameters (Å, °)

O1—C9	1.216 (4)	C1—C12	1.490 (3)
O2—C9	1.359 (3)	C2—C3	1.379 (4)
O2—C10	1.433 (4)	C2—C9	1.469 (5)
O3-C12	1.210 (3)	C3—C4	1.512 (5)
O4C12	1.341 (3)	C4C5	1.526 (4)
O4C13	1.447 (3)	C5—C6	1.518 (4)
O5-C16	1.425 (3)	C6—C7	1.532 (5)
O5-C17	1.409 (5)	C7—C8	1.496 (4)
N1-C3	1.359 (4)	C10-C11	1.481 (4)
N1-C15	1.451 (5)	C13-C14	1.482 (4)
N1-C18	1.459 (3)	C15—C16	1.500 (4)
C1—C2	1.471 (4)	C17—C18	1.503 (4)
C1—C8	1.336 (5)		
C9-02-C10	116.5 (2)	C4—C5—C6	114.1 (2)
C12-04-C13	116.6 (2)	C5—C6—C7	114.2 (3)
C16-05-C17	108.8 (3)	C6—C7—C8	114.4 (2)
C3-N1-C15	123.4 (2)	C1-C8-C7	127.3 (3)
C3-N1-C18	123.8 (3)	O1—C9—O2	121.8 (3)
C15-N1-C18	112.1 (3)	O1-C9-C2	127.0 (3)
C2-C1-C8	124.3 (2)	O2—C9—C2	111.1 (3)
C2-C1-C12	116.3 (3)	O2-C10-C11	107.0 (3)
C8-C1-C12	119.3 (3)	O3-C12-O4	122.7 (2)
C1-C2-C3	121.0 (3)	O3-C12-C1	124.5 (2)
C1-C2-C9	116.5 (2)	O4C12C1	112.8 (2)
C3—C2—C9	122.4 (3)	O4-C13-C14	107.2 (2)
N1-C3-C2	123.5 (3)	N1-C15-C16	111.0(3)
N1-C3-C4	116.6 (2)	O5-C16-C15	112.5 (2)
C2—C3—C4	119.9 (3)	O5-C17-C18	112.0(3)
C3-C4-C5	115.0 (2)	N1-C18-C17	111.3 (2)

All H atoms were geometrically located 0.95 Å from their parent atoms and included using a riding model; displacement parameters were fixed at $1.3U_{eq}$ of the parent atom.

Data collection: CAD-4 Express (Enraf-Nonius, 1993). Data reduction: MolEN (Fair, 1990). Program(s) used to solve structure: SIMPEL in MolEN. Program(s) used to refine structure: LSFM in MolEN. Molecular graphics: ORTEPII (Johnson, 1976) in MolEN. Software used to prepare material for publication: MolEN.

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Lists of atomic coordinates, displacement parameters, structure factors and complete geometry have been deposited with the IUCr (Reference: AB1416). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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2-Amino-6-fluorobenzothiazole

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Abstract

Crystals of 2-amino-6-fluoro-1,3-benzothiazole, C_7H_5 -FN₂S, have an amphiphilic layer-like structure. Each amino substituent donates two protons to hydrogen bonds and accepts one. The ring N atoms accept one proton. The F atoms are not involved in any hydrogen bonds.

Comment

We have been interested in the possibility that covalently bound F atoms might act as hydrogen-bond acceptors. The structure determination of the title compound, (I), was undertaken because it is an uncharged fluorinated organic molecule having hydrogen-bond donors.

† Deceased.